Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Maciej Makowski,^a Marek Lisowski,^b* Anna Maciąg^b and Tadeusz Lis^b

^aInstitute of Chemistry, University of Opole, 48 Oleska Street, 45-052 Opole, Poland, and ^bFaculty of Chemistry, University of Wrocław, 14 Joliot-Curie Street, 50-383 Wrocław, Poland

Correspondence e-mail: mbl@wchuwr.chem.uni.wroc.pl

Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.012 Å Disorder in main residue R factor = 0.072 wR factor = 0.114 Data-to-parameter ratio = 9.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2006 International Union of Crystallography Printed in Great Britain – all rights reserved

N-[*tert*-Butoxycarbonylglycyl-(*Z*)- α , β -dehydrophenylalanylglycyl-(*E*)- α , β -dehydrophenylalanyl]glycine methyl ester dihydrate

The title pentapeptide, $Boc^0 - Gly^1 - \Delta^Z Phe^2 - Gly^3 - \Delta^E Phe^4 - Gly^5 - OMe$, $C_{30}H_{35}N_5O_8 \cdot 2H_2O$, adopts the type I β -turn conformation for the $\Delta^Z Phe^2 - Gly^3$ residues. It is stabilized by a $4 \rightarrow 1$ intramolecular hydrogen bond between the $\Delta^E Phe^4$ NH and Gly¹ CO groups. All the amino acid residues in the pentapeptide sequence are linked *trans* to each other. The crystal structure is stabilized by intra- and intermolecular hydrogen bonds.

Received 3 January 2006 Accepted 18 January 2006 Online 27 January 2006

Comment

 α,β -Dehydroamino acid residues contain a double bond between the C α and C β atoms. They have been found in several microbial peptides and antibiotics (Noda et al., 1983; Spatola, 1983). Dehydropeptides [peptides containing dehydroamino acid residue(s) in their sequences] show enhanced resistance to enzymatic degradation (Shimohigashi et al., 1987). The insertion of dehydroamino acid residues into the peptide sequence also results in a distinct increase of the binding ability of dehydropeptides to metal ions (Brasuń et al., 2004). The dehydroamino acid residues restrict the conformation of the peptide backbone in dehydropeptides and they are strong inducers of folded conformations (Singh & Kaur, 1996). The most studied dehydroamino acid residue so far has been dehydrophenylalanine, Δ^{Z} Phe (Vijayaraghavan *et al.*, 1998, and references therein; Siddiqui, 1999; Kubica et al., 2000). Crystal structures of different Δ^{Z} Phe-containing peptides have shown that Δ^{Z} Phe induces β -turns in short sequences with one Δ^{Z} Phe (Główka *et al.*, 1987; Główka, 1988; Aubry et al., 1991) and a right-handed 310-helical conformation in longer peptides (Rajashankar et al., 1992; Padmanabhan & Singh, 1993; Rajashankar, Ramakumar, Jain & Chauhan, 1995; Rajashankar, Ramakumar, Mal, Jain & Chauhan, 1995; Jain et al., 1997). It has been found (Vijayaraghavan et al., 1998, and references therein) that a Δ Phe residue adopts one of the three conformations with average φ and ψ torsion angles of 80 and 0° , -60 and 140°, or -60 and -30°, or their enantiomeric values. It has also been observed that ΔPhe residues at the (i+2) position in a three-peptide unit sequence induce a type II β -turn conformation with φ and ψ torsion angles values falling close to 80° and 0°, respectively (Singh et al., 1987; Główka, 1988; Patel et al., 1990; Busseti et al., 1992). Studies of sequences containing more than one Δ Phe residue, or one Δ Phe and another dehydroamino acid residue, separated by one or more saturated residue(s), have shown that these peptides adopt a 3_{10} -helical conformation with φ and ψ torsion angles of about -60 and -30° , respectively (Singh & Kaur, 1996; Padyana et al., 2003; Goel et al., 2005). The present paper describes the crystal structure of the title hydrated pentapeptide, (I), containing two dehydrophenylalanyl residues, surrounded by flexible glycyl ones.



The first and second dehydrophenylalanine residues are of the Z and E configuration, respectively. There is one molecule in the asymmetric unit. A perspective view with the numbering scheme is shown in Fig. 1. Table 1 lists selected geometric parameters. The bond lengths of $C\alpha = C\beta$ (C8=C9 and C19=C20) are consistent with those obtained from other studies of pentapeptides containing Δ Phe residues, and correspond well to the classical C=C double-bond distance of 1.337 Å (Dickerson & Geis, 1969). The lengths of the N2-C8, C8-C16, N4-C19 and C19-C27 bonds indicate that in both Δ Phe residues the carbonyl group and N atom are conjugated with the styrene substituent.

Owing to the shortening of the distance between the C α and $C\beta$ atoms, the side-chain atoms of the Δ Phe residues are closer to the main chain than in their saturated counterpart. This results in some rearrangement of the bond angles at $C\alpha$ and C β . The values of the N-C α -C' bond angle in the dehydro residues are found to be smaller than the standard trigonal bond angle of 120° , whereas the bond angles C α - $C\beta - C\gamma$ are considerably larger in both Δ Phe residues. The bond angle N-C α =C β is noticeably larger in the case of the Δ^{Z} Phe² residue, whereas in the case of Δ^{E} Phe⁴ the value of this angle falls below 120°. Constraints imposed on the pentapeptide backbone are partially relaxed through the distortions in the geometry. All the amino acid residues are linked trans to each other. The deviations from the ideal value of $\pm 180^{\circ}$ are not larger than 6°. The values of torsion angles χ^2 $= -4.2 (14)^{\circ}, \chi^{2,1} = -44.9 (13)^{\circ}$ and $\chi^{2,2} = 138.0 (9)^{\circ}$ indicate that the side chain of the Δ^{Z} Phe² residue is synperiplanar, and the $\chi^{4} = 171.5 \ (8)^{\circ}, \ \chi^{4,1} = 167.9 \ (9)^{\circ}$ and $\chi^{4,2} = -10.1 \ (16)^{\circ}$ torsion angles suggest that the side chain of Δ^{E} Phe⁴ is almost planar. The dihedral angles between the C=C and C=O bonds of Δ^{Z} Phe² and Δ^{E} Phe⁴ are -47.0 (10) and -41.1 (15)°, respectively.

There is one intramolecular $4 \rightarrow 1$ hydrogen bond (N4– H4···O3) which indicates the presence of a β -turn conformation for the Δ^{Z} Phe²–Gly³ residues. The torsion angles φ and ψ in the two residues correspond to the type I β -turn. The standard values of φ and ψ angles for that β -turn are -60 and

 -30° , and -90 and 0° . The angles present in the Δ^{Z} Phe² and Glv³ residues are -40 and -42° , and -80 and 0° , respectively. The peptide studied is very similar to $Boc^0-Gly^1-\Delta^ZPhe^2 \text{Gly}^3 - \Delta^E \text{Phe}^4 - \text{L} - \text{Phe}^5 - p - \text{NA}$ (p-NA is p-nitroaniline), whose crystal structure was established a short time ago (Makowski et al., 2005), the only difference being the amino acid residue at position 5 and the C-terminal blocking group. In that peptide, there is a type I' β -turn on the Δ^{Z} Phe² and Gly³ residues and a type II' β -turn on the Δ^E Phe⁴ and Phe⁵ residues. Both these turns are stabilized by $4 \rightarrow 1$ hydrogen bonds, one between Δ^{E} Phe⁴ NH and Gly¹ CO and another between p-NA NH and Gly³ CO. The latter turn is not possible in the case of the title peptide because there is no proton donor at the corresponding position in the peptide chain. As can be seen, the lack of the Phe⁵ residue and the p-NA group results in a β -turn on the N-terminal tetrapeptide which is a mirror image of that in the chiral peptide (Makowski et al., 2005). The conformation of the peptide is stabilized by interand intramolecular hydrogen bonds of different types, namely $N-H\cdots O$, $O-H\cdots O$, $C-H\cdots O$ and $N-H\cdots N$. Geometric parameters of hydrogen bonds are presented in Table 2. The results clearly show that introduction of two α,β -dehydrophenylalanyl residues into the peptide sequence induces the β turn conformation.

Experimental

The synthesis of the title compound was described by Brasuń *et al.* (2004). The crystals were grown by slow diffusion of hexane into an ethyl acetate–methanol (20:1 ν/ν) solution of the compound at room temperature.

Crystal data

$C_{30}H_{35}N_5O_8 \cdot 2H_2O$	Cu $K\alpha$ radiation
$M_r = 629.66$	Cell parameters from 5475
Orthorhombic, $P2_12_12$	reflections
a = 15.371 (4) Å	$\theta = 3-73^{\circ}$
p = 23.889 (6) Å	$\mu = 0.81 \text{ mm}^{-1}$
r = 8.940 (3) Å	T = 100 (2) K
$V = 3282.7 (16) \text{ Å}^3$	Needle, colourless
Z = 4	$0.04 \times 0.03 \times 0.14 \text{ mm}$
$D_x = 1.274 \text{ Mg m}^{-3}$	

Data collection

Xcalibur PX κ -geometry CCD diffractometer ω and φ scans Absorption correction: analytical (*CrysAlis RED*; Oxford Diffraction, 2003) $T_{\min} = 0.910, T_{\max} = 0.983$ 22419 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.072$ $wR(F^2) = 0.114$ S = 1.183775 reflections 417 parameters 3775 independent reflections 1835 reflections with $I > 2\sigma(I)$ $R_{int} = 0.146$ $\theta_{max} = 76.7^{\circ}$ $h = -19 \rightarrow 15$

 $k = -30 \rightarrow 22$

 $l=-9\rightarrow 10$

H atoms treated by a mixture of independent and constrained refinement $w = 1/\sigma^2(F_o^2)$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.40 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.32 \text{ e} \text{ Å}^{-3}$



Figure 1

The molecular structure of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii. Dashed lines indicate intramolecular hydrogen bonds. The disordered methyl group in Boc^0 is drawn with dashed lines.

Table 1

Selected geometric parameters (Å, °).

N2-C8	1.437 (8)	N4-C19	1.430 (9)
C8-C9	1.321 (9)	C19-C20	1.332 (10)
C9-C8-N2	123.8 (7)	C20-C19-N4	116.1 (7)
C9-C8-C16	119.4 (7)	C20-C19-C27	127.5 (8)
N2-C8-C16	115.9 (7)	N4-C19-C27	116.4 (7)
C8-C9-C10	128.0 (7)	C19-C20-C21	132.4 (8)
C1-O1-C5-N1	-177.4 (8)	C16-N3-C17-C18	-80.2(9)
O1-C5-N1-C6	176.3 (8)	N3-C17-C18-N4	0.2 (10)
C5-N1-C6-C7	135.2 (9)	C17-C18-N4-C19	-176.0(6)
N1-C6-C7-N2	-161.0(7)	C18-N4-C19-C27	-52.9 (11)
C6-C7-N2-C8	177.4 (7)	C20-C19-C27-O6	-41.1 (15)
C7-N2-C8-C16	-39.2(11)	N4-C19-C27-N5	-39.9 (11)
C9-C8-C16-O4	-47.0(12)	N4-C19-C20-C21	171.5 (8)
N2-C8-C16-N3	-42.1(10)	C19-C20-C21-C26	-10.1(16)
N2-C8-C9-C10	-4.2(14)	C19-C20-C21-C22	167.9 (9)
C8-C9-C10-C15	138.0 (9)	C19-C27-N5-C28	176.8 (7)
C8-C9-C10-C11	-44.9 (13)		
C8-C16-N3-C17	-174.9 (7)		

Table 2			
Hydrogen-bonding geometry	(Å,	°).	

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdot \cdot \cdot A$
N4-H4···O3	0.88	2.03	2.869 (8)	160
$N3-H3\cdots O6^{i}$	0.88	2.08	2.944 (8)	166
$N2-H2\cdots O5^{i}$	0.88	2.01	2.885 (7)	174
N5-H5···O9	0.88	2.06	2.853 (9)	149
$O9-H9A\cdots O4$	0.86	2.09	2.831 (9)	143
$O9-H9B\cdots O10^{ii}$	0.86	2.07	2.794 (9)	141
O10−H10A···O7 ⁱⁱⁱ	0.86	2.16	2.967 (9)	156
O10−H10B···O5	0.86	2.09	2.947 (9)	177
$C2-H2B\cdots O2$	0.98	2.35	2.970 (10)	121
C3-H3C···O2	0.98	2.42	3.031 (12)	120
$C6-H6A\cdots O2$	0.99	2.36	2.761 (9)	103
$N4-H4\cdots N3$	0.88	2.31	2.741 (8)	110
$N5-H5\cdots N4$	0.88	2.53	2.825 (8)	100
$C26-H26A\cdots O6$	0.95	2.48	3.101 (10)	122
$C28 - H28A \cdots O2^{iv}$	0.99	2.22	3.145 (11)	155

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C28-H28B\cdots O8^{v}$	0.99	2.60	3.337 (10)	131
$C28 - H28A \cdots O2^{iv}$	0.99	2.22	3.145 (11)	155
$C26 - H26A \cdots O6$	0.95	2.48	3.101 (10)	122

Symmetry codes: (i) $\frac{1}{2} + x$, $\frac{1}{2} - y$, -z; (ii) x, y, z - 1; (iii) x, y, 1 + z; (iv) $x - \frac{1}{2}$, $\frac{1}{2} - y$, -z; (v) 1 - x, -y, z.

All H atoms, except those belonging to water molecules, were positioned geometrically, with C–H distances in the range 0.95–0.99 Å and N–H distances of 0.88 Å, and refined using a riding model, with $U_{iso}(H) = 1.5U_{eq}$ (methyl C) or $1.2U_{eq}$ (other C). Water H atoms were located in a difference Fourier map and refined with the constraint O–H = 0.86 Å. In the absence of significant anomalous dispersion effects, Friedel pairs were merged. During refinement, it was found that the position of the C4 atom in the Boc⁰ group was disordered, as a result of the large anisotropic displacement parameter between two alternative sites, and both seemed to have equal occupancies of 0.5 for C4A and C4B. The refinement was carried out using anisotropic displacement parameters for all non-H atoms, except that C4A and C4B were refined isotropically. Due to the small dimensions of the crystal, a large number of reflections have the intensity < $2\sigma(I)$; this results in the high value of R_{int} .

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2003); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2003); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXL97*.

References

- Aubry, A., Pietrzyński, G., Rzeszotarska, B., Boussard, G. & Marraud, M. (1991). Int. J. Peptide Protein Res. 37, 39–45.
- Brasuń, J., Makowski, M., Kołodziej, S. & Swiątek-Kozłowska, J. (2004). J. Inorg. Biochem. 98, 1391–1398.
- Bruker (1997). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Busseti, V., Crisma, M., Toniolo, C., Salvadari, S. & Balboni, G. (1992). Int. J. Biol. Macromol. 14, 23–28.
- Dickerson, R. E. & Geis, I. (1969). *The Structure and Action of Proteins*, p. 13. New York: Harper and Row.
- Główka, M. L. (1988). Acta Cryst. C44, 1639-1641.
- Główka, M. L., Gilli, G., Bertolasi, V. & Makowski, M. (1987). Acta Cryst. C43, 1403–1406.
- Goel, V. K., Somvanshi, R. K., Dey, S. & Singh, T. P. (2005). J. Peptide Res. 66, 68–74.
- Jain, R. M., Rajashankar, K. R., Ramakumar, S. & Chauhan, V. S. (1997). J. Am. Chem. Soc. 119, 3205–3211.
- Kubica, Z., Koźlecki, T. & Rzeszotarska, B. (2000). Chem. Pharm. Bull. 48, 296–297.
- Makowski, M., Brzuszkiewicz, A., Lisowski, M. & Lis, T. (2005). *Acta Cryst.* C**61**, 0424–0426.
- Noda, K., Shimohigasi, Y. & Izumiya, N. (1983). Peptides: Analysis, Synthesis, Biology, Vol. 5, edited by E. Gross & J. Meienhofer, pp. 285–339. New York: Academic Press.
- Oxford Diffraction (2003). CrysAlis CCD and CrysAlis RED. Version 1.171. Oxford Diffraction Poland, Wrocław, Poland.
- Padmanabhan, B. & Singh, T. P. (1993). Biopolymers, 33, 613-619.
- Padyana, A. K., Ramakumar, S., Mathur, P., Jagannathan, N. R. & Chauhan, V. S. (2003). J. Peptide Sci. 9, 54–63.
- Patel, H. C., Singh, T. P., Chauhan, V. S. & Kaur, P. (1990). Biopolymers, 29, 509–515.
- Rajashankar, K. R., Ramakumar, S. & Chauhan, V. S. (1992). J. Am. Chem. Soc. 114, 9225–9226.
- Rajashankar, K. R., Ramakumar, S., Jain, R. M. & Chauhan, V. S. (1995). J. Am. Chem. Soc. 117, 11773–11779.
- Rajashankar, K. R., Ramakumar, S., Mal, T. K., Jain, R. M. & Chauhan, V. S. (1995). *Biopolymers*, 35, 141–147.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Gottingen, Germany.

- Shimohigashi, Y., Kodama, H., Imazu, S., Horimoto, H., Sagakuchi, K., Waki, M., Uchida, H., Kondo, M., Kato, T. & Izumiya, N. (1987). FEBS Lett. 222, 251–255.
- Siddiqui, M. Z. (1999). Int. J. Biol. Macromol. 26, 17-21.
- Singh, T. P., Haridas, M., Chauhan, V. S., Kumar, A. & Witerbo, D. (1987). Biopolymers, 26, 819–829.

Singh, T. P. & Kaur, P. (1996). Prog. Biophys. Mol. Biol. 66, 141-165.

- Spatola, A. F. (1983). Chemistry and Biochemistry of Amino Acids, Peptides and Proteins, Vol. VII, edited by B. Weinstein, pp. 267–357. New York: Marcel Dekker.
- Vijayaraghavan, R., Kumar, P., Dey, S. & Singh, T. P. (1998). J. Peptide Res. 52, 89–94.